

Environmental Geotechniques

1) Aroclor 1260 $\log K_{ow} = 6.91$ is a PCB

$$\begin{aligned} \text{For PCBs } \log K_{oc} &= 0.544 \log K_{ow} + 1.377 \\ &= 0.544 (6.91) + 1.377 \\ &= 5.14 \end{aligned}$$

$$\begin{aligned} K_{oc} &= 138,038 \\ K_{ow} &= 8,128,305 \end{aligned}$$

$$f^*_{oc} = \frac{SSA}{200 (K_{ow})^{0.84}} = \frac{35}{200 (8,128,305)^{0.84}} = 2.75 \times 10^{-7}$$

$$f_{oc} = 0.25\% = 0.0025$$

$f_{oc} > f^*_{oc}$ so equation is valid.

$$\begin{aligned} K_d &= K_{oc} \times f_{oc} \\ &= 138,038 \times 0.0025 \\ &= 345.1 \end{aligned}$$

$$R = 1 + \frac{\rho_d K_d}{n} = 1 + \frac{1.55 \text{ g/cm}^3 \times 345.1 \text{ mL/g}}{0.45}$$

$$= 1189.7 \rightarrow \text{GW velocity } 1189.7 \text{ times faster than rate of transport of Aroclor 1260}$$

2) With advection dominant: plug flow, so full concentration

$$L = \frac{1.0 + 0.3}{1} = 1.3 \dots$$

$$v_s = \frac{K_i}{n_e} = \frac{1 \times 10^{-9} \text{ m/s} \times 1.3}{0.45} = 2.89 \times 10^{-9} \text{ m/s} \dots$$

$$t = \frac{1.0 \text{ m}}{2.89 \times 10^{-9} \text{ m/s}} = 3.46 \times 10^8 \text{ s} = 11.0 \text{ years}$$

3) With diffusion dominant:

$$\frac{c}{c_0} = \text{erfc}\left(\frac{x}{2\sqrt{\frac{D^*t}{R}}}\right) = 0.25 = \text{erfc}(\beta) \dots$$

$$\beta = 0.8 \text{ gives } \text{erfc}(\beta) = 0.258$$

$$\beta = 0.85 \text{ gives } \text{erfc}(\beta) = 0.229$$

$$0.258 - 0.229 = 0.029$$

$$\frac{0.008}{0.029} \times 0.05 = 0.014$$

$$0.25 = \text{erfc}(0.814) \dots$$

$$0.814 = \frac{1 \text{ m}}{2\sqrt{\frac{1.2 \times 10^{-10} \text{ m}^2/\text{s} \times t}{1189.7}}}$$

$$D^* = 1.2 \times 10^{-10} \frac{\text{m}^2}{\text{s}} \times \frac{3600 \text{ s}}{\text{hr}} \times \frac{24 \text{ hr}}{\text{d}} \times \frac{365 \text{ d}}{\text{yr}} = 0.00378 \frac{\text{m}^2}{\text{yr}}$$

$$0.814 = \frac{1 \text{ m}}{2\sqrt{\frac{0.00378 \text{ m}^2/\text{yr} \times t}{1189.7}}}$$

$$t = \frac{1 \text{ m}}{4 \times \frac{0.00378}{1189.7} \times 0.814 \times 0.814} = \frac{1}{0.00008421}$$

$$= 118,750 \text{ years. } \dots$$

4) With advection and diffusion both important

$$\frac{C}{C_0} = 0.25 = \frac{1}{2} \left\{ \operatorname{erfc} \left(\frac{Rz - V_s t}{2\sqrt{D^* R t}} \right) + \exp \left(\frac{V_s z}{D^*} \right) \operatorname{erfc} \left(\frac{Rz + V_s t}{2\sqrt{D^* R t}} \right) \right\}$$

$$\frac{V_s z}{D^*}$$

$$V_s = 2.89 \times 10^{-9} \frac{\text{m}}{\text{s}} \times 3600 \frac{\text{s}}{\text{hr}} \times 24 \frac{\text{hr}}{\text{d}} \times 365 \frac{\text{d}}{\text{yr}} = 0.0911 \frac{\text{m}}{\text{yr}}$$

$$\frac{V_s z}{D^*} = \frac{0.0911 \text{ m/yr} \times 1 \text{ m}}{0.00378 \text{ m}^2/\text{yr}} = 24.1$$

$$\exp(24.1) = 2.93 \times 10^{10}$$

$$\left. \begin{array}{l} R = 1189.7 \\ D^* = 0.00378 \\ V_s = 0.0911 \end{array} \right\} \text{ consider } \operatorname{erfc} \left(\frac{Rz + V_s t}{2\sqrt{D^* R t}} \right) = \operatorname{erfc}(\beta)$$

$$\begin{aligned} \text{at } t = 10 \text{ years} \quad \beta &= \frac{1189.7 \times 10 + 0.0911 \times 10}{2\sqrt{0.00378 \times 1189.7 \times 10}} \\ &= \frac{11897 + 0.911}{13.4} = 887. \end{aligned}$$

β is too large and will always be too large so the second term will always be equal to zero, so ignore the second term.

$$\frac{C}{C_0} = 0.25 = \frac{1}{2} \operatorname{erfc} \left(\frac{Rz - V_s t}{2\sqrt{D^* R t}} \right)$$

$$0.5 = \operatorname{erfc}(\beta); \quad \beta \text{ between } 0.45 \text{ and } 0.5$$

4) cont'd. $\beta = 0.45$ $\text{erfc}(\beta) = 0.524$ $0.524 - 0.480 = 0.044$
 $\beta = 0.50$ $\text{erfc}(\beta) = 0.480$ $\frac{0.24 \times 0.05}{0.44} = 0.027$
 $\beta = 0.477$ $\text{erfc}(\beta) = 0.50$

$$\frac{Rx - \sqrt{st}}{2\sqrt{D \cdot R \cdot t}} = \frac{1189.7 \times 1 - 0.0911t}{2\sqrt{0.00378 \times 1189.7 \times t}} = 0.477$$

$$1189.7 - 0.0911t = 2.023\sqrt{t}$$

$$1189.7 - 2.023\sqrt{t} - 0.0911t = 0$$

$$1189.7 - 202.3 - 911 = 76.4$$

$$\text{by } t = 10,000 \text{ years}$$

$$\text{by trial and error } t = 10,756 \text{ years.}$$

5) Peclet number = $\frac{V_s d}{D_0}$

for a clay soil the grain size may be from 0.001 to 0.005 mm according to ASTM D 4122 "Geoenvironmental Engineering", Sharma and Reddy, 2004, p. 46.

use $d = 0.0025 \text{ mm} = 2.5 \times 10^{-6} \text{ m}$.
 $V_s = 2.89 \times 10^{-9} \text{ m/s}$

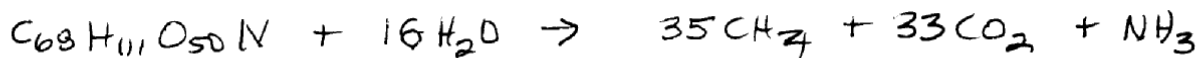
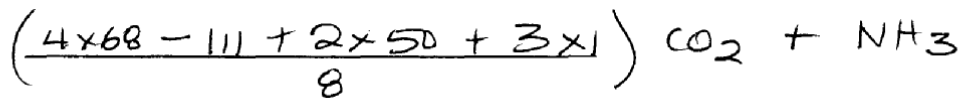
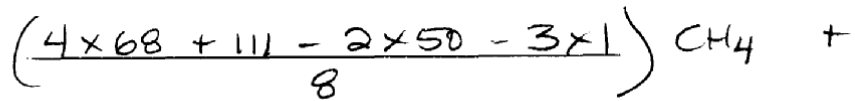
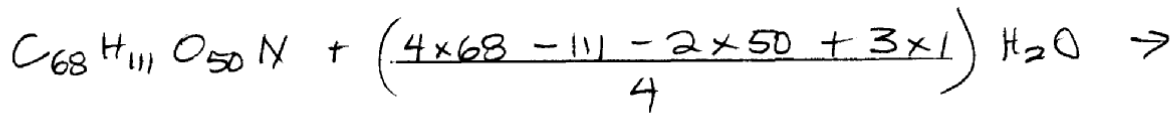
$$D_0 = \frac{D^*}{\tau} \text{ where } \tau \text{ ranges from } 0.01 \text{ to } 2.5 \text{ let } \tau = 0.2$$

$$D_0 \approx \frac{1.2 \times 10^{-10} \text{ m}^2/\text{s}}{0.2} = 6 \times 10^{-10} \text{ m}^2/\text{s}$$

$$\frac{V_s d}{D_0} = \frac{2.89 \times 10^{-9} \text{ m/s} \times 2.5 \times 10^{-6} \text{ m}}{6 \times 10^{-10} \text{ m}^2/\text{s}} = 1.20 \times 10^{-5}$$

so diffusion dominates

6) Waste is represented by $C_{68}H_{111}O_{50}N$



68×12	816	$16 \times (2+16)$	$35(12+4)$	$33(12+32)$	$14+3$
111×1	111				
50×16	800				
1×14	<u>14</u>				
	1741	288	560	1,452	17

1000 kg

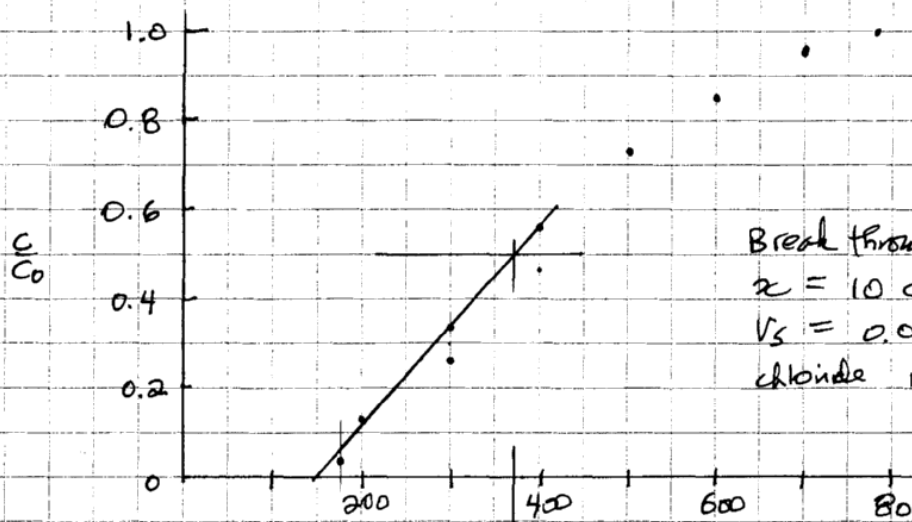
$$0.7177 \frac{kg}{m^3}$$

$$\frac{1000 \text{ kg}}{1741} = \frac{x \text{ kg}}{560}$$

$$CH_4 = 321.65 \text{ kg}$$

$$\frac{321.65 \text{ kg}}{0.7177 \frac{kg}{m^3}} = 448 \text{ m}^3$$

$$= 448,000 \text{ L } CH_4$$



Break through at 370 hrs.
 $x = 10$ cm
 $v_s = 0.025$ cm/hr
 chloride non-reactive, $R=1$

$$\frac{c}{c_0} = 0.5 = 0.5 \left\{ \operatorname{erfc} \left(\frac{Rx - v_s t}{2\sqrt{D^* R t}} \right) + \exp \left(\frac{v_s x}{D^*} \right) \operatorname{erfc} \left(\frac{Rx + v_s t}{2\sqrt{D^* R t}} \right) \right\}$$

$$1 = \left\{ \operatorname{erfc} \left(\frac{10 \text{ cm} - 0.025 \text{ cm/hr} \times 370 \text{ hr}}{2\sqrt{D^*} \sqrt{370 \text{ hr}}} \right) + \exp \left(\frac{0.025 \text{ cm/hr} \times 10 \text{ cm}}{D^*} \right) \right\}$$

$$1 = \left\{ \operatorname{erfc} \left(\frac{10 \text{ cm} - 9.25 \text{ cm}}{38.47\sqrt{D^*}} \right) + \exp \left(\frac{0.25 \text{ cm}^2/\text{hr}}{D^*} \right) \operatorname{erfc} \left(\frac{10 + 9.25}{38.47\sqrt{D^*}} \right) \right\}$$

$$\text{Try } D^* = 2.75 \times 10^{-2} \text{ cm}^2/\text{hr}$$

$$\begin{aligned} & \operatorname{erfc}(0.1176) + \exp(9.09) \operatorname{erfc}(3.02) \\ & \quad + 8866 \times 0.000022 \\ & \quad 0.86 \quad \quad 0.95 \end{aligned}$$

$$\begin{array}{r} 0.15 \quad 1.76 = z \quad 0.887 \\ -0.12 \quad 5. \quad 0.005 \quad 0.832 \\ \hline 0.05 \quad 2 = 0.019 \quad 0.055 \\ \quad \quad -0.887 \\ \quad \quad 0.868 \end{array}$$

$$= 1.06$$

$$\text{Try } D^* = 2.8 \times 10^{-2} \text{ cm}^2/\text{hr}$$

$$\begin{aligned} & \operatorname{erfc}(0.1165) + \exp(8.93) \operatorname{erfc}(2.99) \\ & \quad 0. \quad \quad 7555 \times 0.000039 \\ & \quad \quad \quad 0.295 \end{aligned}$$

$$1.20$$

$$\text{Try } D^* = 2.7 \times 10^{-2} \text{ cm}^2/\text{hr}$$

$$\begin{aligned} & \operatorname{erfc}(0.119) + \exp(9.26) \operatorname{erfc}(3.04) \\ & \quad 0.86 \quad \quad 10,509 \times 0.000027 \\ & \quad \quad \quad 0.18 \end{aligned}$$

$$=$$

$$= 1.04$$

$$2.7 \times 10^{-2} \frac{\text{cm}^2}{\text{hr}} \times \frac{\text{m}^2}{10^4 \text{cm}^2} \times \frac{\text{hr}}{3600 \text{sec}} = 7.5 \times 10^{-10} \frac{\text{m}^2}{\text{sec}}$$

This is reasonable because

$$D_0 z = D^*$$

$$\text{if } D_0 = 20.3 \times 10^{-10} \frac{\text{m}^2}{\text{s}} \text{ and } z = 0.37$$

then you get this value for D^* so this is
reasonable since $z = 0.01$ to 0.5 usually.